

**School of Engineering and Computer Science**  
**Mech 301: Thermodynamics**

<b>Catalog Data:</b>	<b>Mech 301 Thermodynamics;</b> 3 credits Thermodynamic properties of matter, ideal and real gases, work and heat, first and second laws and their application to engineering systems.
<b>Class Schedule:</b>	Three 50-minute lectures per week, for one semester.
<b>Laboratory Schedule:</b>	None
<b>Prerequisites by Course:</b>	Phys 201; Recommended prerequisite: Math 220, Math 315
<b>Prerequisites by Topic:</b>	<ol style="list-style-type: none"> <li>1. Basic knowledge of Physics.</li> <li>2. Good understanding of Calculus.</li> </ol>
<b>Required Texts:</b>	Fundamentals of Engineering Thermodynamics, M.J. Moran, H.N. Shapiro
<b>Elective or Required Course:</b>	Required
<b>Course Coordinator:</b>	Dr. Amir Jokar
<b>Course Objectives:</b>	<ol style="list-style-type: none"> <li>1. Ability to explain the basic principles of classical thermodynamics.</li> <li>2. Ability to find the thermodynamic properties of pure substances (using tables, charts, and ideal gas law) and apply them to thermodynamic analysis.</li> <li>3. Ability to identify, formulate, and solve engineering problems of closed and open systems involving heat and work interactions with the surroundings.</li> <li>4. Ability to apply the First and Second Laws of Thermodynamics to analyze thermal systems.</li> <li>5. Ability to determine properties of ideal gas mixtures and apply them to thermodynamic analysis.</li> <li>6. Ability to calculate properties of air-water mixtures and apply them to thermodynamic analysis.</li> </ol>
<b>Topics Covered:</b>	<ol style="list-style-type: none"> <li>1. Thermodynamic Concepts</li> <li>2. Closed Systems- 1st Law</li> <li>3. Properties of pure substances</li> <li>4. Open Systems- 1st Law</li> <li>5. Systems- 2nd Law</li> <li>6. Thermodynamic Cycles</li> <li>7. Ideal Gas Mixtures</li> <li>8. Thermodynamics of state</li> </ol>
<b>Lab Experiments and Activities:</b>	None
<b>Course Outcomes:</b>	Students will be able to: A-1. Apply mathematical principles to derive equations for conservation of mass, momentum, and energy for the thermodynamic problems. A-2. Choose the closed or open system analysis to solve for the mass, momentum and energy balance of the systems. A-3. Review the conservation laws; mass, momentum, and energy. A-4. Analyze 1 <sup>st</sup> and 2 <sup>nd</sup> law of thermodynamics, thermodynamic cycles, ideal

	<p>gas, and psychometrics on the engineering processes and systems.</p> <p>E-1. Classify the thermodynamic problems, such as based on the closed or open system, depending on conditions or assumptions defined in the engineering problems.</p> <p>E-2. Examine different methods, such as integral or differential forms, for specific thermodynamic problems.</p> <p>E-3. Use analytical, computational, or experimental analysis to obtain solutions for a specific thermodynamic problem.</p> <p>H-3. Evaluate the interaction between a system and its surroundings, and the effect of system on the environment and universe.</p> <p>J-1. Recognize the impact of environmental issues based on thermodynamic laws.</p> <p>J-2. Apply thermodynamic laws to describe engineering decisions for new energy resources.</p> <p>K-4. Numerically analyze the thermodynamic problems.</p>		
<b>Required or Elective Course:</b>	Required		
<b>Contribution to Professional Component:</b>	Engineering Topics		
<b>Relationship of Course to Program:</b>	Meets: Educational Objectives <u>1, 2, 3, 4</u> Program Outcomes <u>A, E, H, J, K</u>		
<b>Prepared by:</b>	Amir Jokar	Date:	November 1, 2006
<b>Approved by CAC:</b>			